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(54) **Tapered block copolymers of monovinylarenes and conjugated dienes**

Blockcopolymerisate von Monovinylarenen und konjugierten Dienen mit Endblöcken von  
kontinuierlich sich ändernder Zusammensetzung

Copolymères à base de diènes conjugués et de monovinylarènes comprenant des blocs à séquences  
progressives

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**EP-A- 0 512 530** **US-A- 4 167 545**

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**Description****Field of the Invention**

5 This invention relates to tapered block copolymers of monovinylarenes and conjugated dienes and methods of preparation of these tapered block copolymers.

**Background of the Invention**

10 There has developed in the polymer field, and especially in the packaging and related industries, a need for thermoplastic polymers that can be formed into colorless, transparent articles having good impact strength and ductility. There are needs for polymers which are useful as single components as well as for use in blends with other commonly used polymers to make articles with improved properties. The polymers satisfying these needs should be suitable for use with conventional extrusion, injection and blow molding equipment and also suitable for use in other methods of forming plastics into containers, tubes, films, fibers, and the like. Polystyrene, high impact polystyrene, branched block copolymers, and the like have been developed to meet these criteria with various degrees of satisfaction.

15 Much effort has been directed to the preparation of substantially transparent block copolymer resins with a variety of block structures produced by a variety of monomer addition sequences and a variety of coupling agents. Desirable properties and an economic advantage can be obtained by blending some monovinylaromatic-conjugated diene copolymers with polystyrene polymers. However, because blueness of blends of monovinylaromatic-conjugated diene copolymers with polystyrene polymers cannot be predicted by rule of mixtures behaviors, getting a desirable combination of properties can be a complicated task. Sometimes relatively colorless monovinylaromatic-conjugated diene copolymers give blends with high blueness when blended with colorless general purpose polystyrene. Exemplary publications in this field are US-A-4 167 545 and EP-A-0 512 530.

20 Blueness of articles formed from various copolymers and blends of copolymers with other polymers is a longstanding problem in applications where colorless materials which also have good impact strength and ductility are desirable. Specific examples include materials for water and food containers.

**Summary of the Invention**

30 It is an object of this invention to provide resinous tapered block copolymers of vinyl-substituted aromatic hydrocarbons and conjugated dienes from which can be made articles with low blueness and acceptable impact strength and ductility.

35 It is another object of this invention to provide resinous tapered block copolymers of vinyl-substituted aromatic hydrocarbons and conjugated dienes suitable for use in blends with polymers of styrene, particularly blends from which can be made articles that exhibit low blueness and acceptable impact strength and/or ductility. Further, it is an object to provide resinous tapered block copolymers of vinyl-substituted aromatic hydrocarbons and conjugated dienes which can be used in smaller amounts than some of the other commonly used resinous polymodal monovinyl substituted aromatic-conjugated diene block copolymers in blends with polymers of styrene to achieve similarly low blueness levels and good impact strength and/or ductility in articles made from the blends.

40 A further object of this invention is to provide novel processes for making resinous tapered block monovinylaromatic/conjugated diene copolymers, including copolymers suitable for use in blends.

In a first embodiment of the invention, copolymers are prepared under solution polymerization conditions in a reaction zone by:

45 (a) charging a monovinylaromatic monomer and an initiator in the presence of a randomizer, which is an ether, a thioether, or a tertiary amine, and allowing polymerization to occur until essentially no free monomer is present; thereafter

(b) charging an initiator and a monovinylaromatic monomer and allowing polymerization to occur until essentially no free monomer is present; thereafter

50 (c) charging a mixture of a monovinylaromatic monomer and a conjugated diene monomer and allowing polymerization to occur until essentially no free monomer is present; and thereafter

(d) optionally charging a mixture of a monovinylaromatic monomer and a conjugated diene monomer; and thereafter

55 (e) charging the reaction mixture with a coupling agent, wherein in steps (a) and (b) 0.005-1.5 parts by weight per 100 parts by weight of total monomers each of initiator, and 20-60 parts by weight per 100 parts by weight of total monomers each of the monovinylaromatic monomer are employed.

Detailed Description of the Invention

We have discovered monovinylaromatic/conjugated diene tapered block copolymers which can be used neat or blended with polymers of styrene to produce resins which can be formed into articles with low blueness and advantageous impact properties and/or ductility.

The polymers of this invention are characterized as resinous non-rubbery block copolymers of at least one conjugated diene with at least one monovinylarene, having at least one random-tapered block and are prepared so that, when the choice of coupling agent permits, at least a portion of the final product is of a branched, coupled character.

The polymers prepared according to this invention contain from 55 to 95, preferably from 60 to 90, more preferably from 65 to 85, weight percent of copolymerized monovinyl aromatic monomer based on the weight of total monomers employed. Correspondingly, the inventive copolymers contain from 45 to 5, preferably from 40 to 10, and more preferably from 35 to 15 weight percent copolymerized conjugated diene monomer based on the total weight of monomers in the copolymer.

The coupled portions of the resinous polymodal block copolymers of this invention have terminal polymonovinylarene blocks on the extending arms of each linear or radial copolymer molecule, and further contain at least one internal tapered block of monovinylarene and conjugated diene. The resinous copolymeric polymodal products also contain portions of linear uncoupled block copolymers of poly(monovinylarene)-poly(conjugated diene); the linear uncoupled block copolymer content is considered to be an important portion of the resinous product with respect to its overall properties.

Components

The process of this invention can be carried out using as an initiator any of the organomonoalkali metal compounds of the formula RM wherein R is an alkyl, cycloalkyl or arylcarbanion containing 4 to 8 carbon atoms and M is an alkali metal cation. Mixtures of organoalkali metal compounds can be used. The presently preferred initiators are alkylmonolithium compounds, especially n-butyllithium or sec-butyllithium.

The conjugated diene monomers which can be used contain 4 to 6 carbon atoms and include 1,3-butadiene, 2-methyl-1,3-butadiene, 2-ethyl-1,3-butadiene, 2,3-dimethyl-1,3-butadiene and 1,3-pentadiene and mixtures thereof. Each of the charges containing conjugated diene in the same sequence of charges may be the same, but is not necessarily the same, conjugated diene monomer or mixture of conjugated diene monomers. The presently preferred conjugated diene monomer is 1,3-butadiene.

The monovinylaromatic monomers which can be used contain 8 to 12 carbon atoms and include styrene, alpha-methylstyrene, 4-methylstyrene, 3-methylstyrene, 2-methylstyrene, 4-ethylstyrene, 3-ethylstyrene, 2-ethylstyrene, 4-tert-butylstyrene, 2,4-dimethylstyrene and condensed aromatics such as vinyl naphthalene and mixtures thereof. Each of the charges containing monovinylaromatic monomer in the same sequence of charges may be the same, but is not necessarily the same, monovinylaromatic monomer or mixture of monovinylaromatic monomers. The presently preferred monovinylaromatic monomer is styrene.

Compounds employed as randomizers according to the invention are ethers, thioethers (sulfides) and tertiary amines. It is usually preferred to use ethers and sulfides in which the radicals attached to the oxygen or sulfur atoms are hydrocarbon radicals. Specific examples of such polar materials include dimethyl ether, diethyl ether, ethyl methyl ether, ethyl propyl ether, di-n-propyl ether, di-n-octyl ether, anisole, dioxane, 1,2-dimethoxyethane, dibenzyl ether, diphenyl ether, 1,2-dimethoxybenzene, tetramethylene oxide (tetrahydrofuran), dimethyl sulfide, diethyl sulfide, di-n-propyl sulfide, di-n-butyl sulfide, methyl ethyl sulfide, dimethylethylamine, tri-n-propylamine, tri-n-butylamine, trimethylamine, triethylamine, tetramethylethylenediamine, tetraethylethylenediamine, N,N-di-methylaniline, N-methyl-N-ethylaniline and N-methylmorpholine. It is to be understood also that mixtures of these polar compounds can be employed in the practice of the present invention. The polar compounds are generally used in admixture with the hydrocarbon diluent. Presently preferred are either tetrahydrofuran or diethyl ether.

Among the suitable coupling agents are the di- or multivinylaromatic compounds, di- or multi epoxides, di- or multiisocyanates, di- or multiimines, di- or multialdehydes, di- or multiketones, alkoxytin compounds, di- or multihalides, particularly silicon halides and halosilanes, mono-, di-, or multianhydrides, mono-, di-, or multiesters, preferably the esters of monoalcohols with polycarboxylic acids, diesters which are esters of monohydric alcohols with dicarboxylic acids, lactones, monobasic acids with polyalcohols such as glycerol, including compounds containing two or more groups and mixtures of two or more compounds.

Examples of suitable vinylaromatic coupling agents include, but are not limited to, divinylbenzene, 1,2,4-trivinylbenzene, 1,3-divinylnaphthalene, 1,3,5-trivinylbenzene, 2,4-divinylbiphenyl and p-diisopropenylbenzene. Of these, the divinylaromatic hydrocarbons are preferred, particularly divinylbenzene in either its ortho, meta, or para isomer. Commercial divinylbenzene which is a mixture of the three isomers and other compounds is satisfactory.

Epoxidized hydrocarbon polymers such as epoxidized liquid polybutadiene and epoxy compounds such as 1,2;

5,6; 9,10-triepoxydecane can be used as coupling agents.

Organoalkyl phosphites and arylalkyl phosphites are considered useful as coupling agents in this invention.

Examples of suitable multiisocyanate coupling agents include benzene-1,2,4-trisocyanate and naphthalene-1,2,5,7-tetraisocyanate. Commercially available products known as PAPI-1, a polyarylpolyisocyanate having an average of 3 isocyanate groups per molecule and an average molecular weight of about 380 are suitable.

The multiimines, also known as multiaziridinyl compounds, such as those containing 3 or more aziridine rings per molecule, are useful as coupling agents. Other compounds useful as coupling agents include tetravinyl silane, trivinyl phosphine, the triaziridinyl phosphine oxides or sulfides such as tri(1-aziridinyl)phosphine oxide, tri(2-methyl-1-aziridinyl)phosphine oxide and tri(2-ethyl-3-decyl-1-aziridinyl)phosphine sulfide.

The multialdehyde coupling agents are represented by compounds such as 1,6,7-naphthalenetetracarboxyaldehyde, 1,7,9-anthracenetetracarboxyaldehyde, 1,3,5-pentanetracarboxyaldehyde, and similar multialdehyde-containing aliphatic and aromatic compounds. The multiketones are represented by compounds such as 1,6,9,10-anthracenetetrone and 2,3-diacetonylcyclohexanone. Examples of the multianhydrides include pyromellitic dianhydride and styrene-maleic anhydride copolymers. Examples of the multiesters include diethyladipate, triethylcitrate, and 1,3,5-benzenetracarboxylic acid triethyl ester.

Among the multihalide coupling agents are the silicon tetrahalides such as silicon tetrachloride, silicon tetrabromide, and silicon tetraiodide; the trihalosilanes such as trichlorosilane, trichloroethylsilane and tribromobenzylsilane; and the multihalogen-substituted hydrocarbons, such as 1,3,5-tri(bromomethyl)benzene and 2,5,6,9-tetrachloro-3,7-decadiene, in which the halogen is attached to a carbon atom which is alpha to an activating group such as an ether linkage, a carbonyl group, or a carbon-to-carbon double bond. Substituents inert with respect to lithium atoms in the terminally reactive polymer can also be present in the active halogen-containing compounds. Alternatively, other suitable reactive groups different from the halogens as described above can be present.

Other metal multihalides, particularly those of tin, lead, or germanium, can be employed as coupling and branching agents. Silicon or other metal multialkoxides, such as silicon tetraethoxide, are also suitable coupling agents.

Examples of compounds containing more than one type of functional group include 1,3-dichloro-2-propanone, 2,2-dibromo-3-decanone, 2,4-dibromo-3-pentanone, 1,2; 4,5-diepoxy-3-pentanone, 1,2; 4,5-diepoxy-3-hexanone, 1,2; 11,12-diepoxy-8-pentadecanone and 1,3; 18,19-diepoxy-7,14-eicosanedione.

Useful multifunctional coupling agents include epoxidized vegetable oils such as epoxidized soybean oil, epoxidized linseed oil and the like or mixtures thereof.

The presently preferred coupling agent is epoxidized vegetable oil. Presently preferred is epoxidized soybean oil.

#### Process

The unique polymodal tapered block character of the polymer and low blueness and good impact strength and/or ductility of articles made from the polymer or blends of the polymer of the first embodiment of this invention are produced by the unique sequences of two separate initial charges of monovinylaromatic monomer and initiator, followed by a separate charge of a mixture of monovinylaromatic monomer and conjugated diene, and a subsequent coupling step.

The unique polymodal tapered block character of the polymer and low blueness and good impact strength and/or ductility of articles made from the polymer or blends of the polymer of the second embodiment of this invention are produced by the process of the first embodiment with the addition of a second separate charge of a mixture of monovinyl aromatic and conjugated diene monomers next preceeding the coupling step.

In each of the two embodiments of this invention the first initiator charge produces active living monovinyl aromatic component polymer blocks with alkali metal atoms (from the initiator) on at least one end to form active reaction sites. Each subsequent monomer charge adds monomer to the living polymer chain at the alkali metal reaction site. At each stage of charging, polymerization is allowed to continue until essentially no free monomer is present.

With each subsequent charge which includes initiator a new polymer-alkali metal species will be produced, and each subsequent monomer charge has an opportunity for polymerization of part of the charge with each of the existing polymer-alkali metal species. Each of the active living polymer chains will have terminal monovinyl aromatic blocks on both ends after polymerization of each monomer charge containing monovinyl aromatic. Even when mixtures of monovinyl aromatic monomer and conjugated diene are charged, the polymer chains will have terminal monovinyl aromatic blocks on both ends. After virtually complete polymerization of the final monomer charge, the active living linear block copolymers are charged with a difunctional or polyfunctional coupling agent to allow coupling of each of the living species with each of the other living species or with others of the same living species to form the desired polymodal tapered block copolymers. If the coupling agent is not 100 percent efficient and/or if less or more than a stoichiometric amount of coupling agent is used, there can be some uncoupled terminated polymer chains of each of the species in the final reaction mixture.

Use of difunctional coupling agents will produce predominantly linear polymer chains. Depending upon functionality, various degrees and kinds of branching may be accomplished with polyfunctional coupling agents. Variations in

the amount of a particular polyfunctional coupling agent also can be used to manipulate the degree and kind of branching at the coupling sites.

The charging sequences of this invention and the resulting polymers at each stage are exemplified using a selected monovinylaromatic monomer, conjugated diene and polyfunctional coupling agent in the following Tables 1 and 2.

TABLE 1

Invention Charging Sequence (First Embodiment)		
Charge	Contents of Charge	Resulting Polymer Chains
(a)	randomizer, initiator <sub>1</sub> and styrene <sub>1</sub>	S <sub>1</sub> -Li <sub>1</sub>
(b)	initiator <sub>2</sub> and styrene <sub>2</sub>	S <sub>1</sub> -S <sub>2</sub> -Li <sub>1</sub> S <sub>2</sub> -Li <sub>2</sub>
(c)	butadiene <sub>1</sub> and styrene <sub>3</sub>	S <sub>1</sub> -S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -Li <sub>1</sub> S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -Li <sub>2</sub>
(e)*	coupling agent	polymodal tapered block copolymers with styrene terminal blocks

where

S = styrene

B = butadiene

B/S = tapered block

Li = residue from a monoalkali metal initiator remaining on the end of the polymerization chain or reaction site prior to termination or coupling.

subscripts = designation of the numerical order in which that particular component was charged or formed.

\* step (d) is an optional step according to claim 1 and is left out in the first embodiment

TABLE 2

Invention Charging Sequence (Second Embodiment)		
Charge	Contents of Charge	Resulting Polymer Chains
(a)	randomizer, initiator <sub>1</sub> and styrene <sub>1</sub>	S <sub>1</sub> -Li <sub>1</sub>
(b)	initiator <sub>2</sub> and styrene <sub>2</sub>	S <sub>1</sub> -S <sub>2</sub> -Li <sub>1</sub> S <sub>2</sub> -Li <sub>2</sub>
(c)	butadiene <sub>1</sub> and styrene <sub>3</sub>	S <sub>1</sub> -S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -Li <sub>1</sub> S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -Li <sub>2</sub>
(d)	butadiene <sub>2</sub> and styrene <sub>4</sub>	S <sub>1</sub> -S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -B <sub>2</sub> /S <sub>4</sub> -Li <sub>1</sub> S <sub>2</sub> -B <sub>1</sub> /S <sub>3</sub> -B <sub>2</sub> /S <sub>4</sub> -Li <sub>2</sub>
(e)	coupling agent	polymodal tapered block copolymers with styrene terminal blocks
where S = styrene B = butadiene B/S = tapered block Li = residue from a monoalkali metal initiator remaining on the end of the polymerization chain or reaction site prior to termination or coupling. subscripts = designation of the numerical order in which that particular component was charged or formed.		

The randomizer is usually added with the diluent initially charged to the reactor. Each of the charges which has two monomers may be either a mixture of the two monomers or simultaneous charging of two separate monomers.

As can be seen from the intermediate products listed in the charging sequence tables above, in two embodiments of the invention there are at least two distinct species of polymer chains before coupling. Thus, polymodal tapered block copolymers comprising relatively high and low molecular weight species can be produced.

Tapered blocks in each of the growing polymer chains are produced by simultaneously charging with at least two monomers as shown in the preceding tables of the inventive charging sequences.

The randomizer regulates tapering or random polymerization of the monovinylaromatic monomer and the conjugated diene in a mixed monomer charge. Choice of randomizer can be used to manipulate the degree and direction of taper in blocks resulting from charges of mixtures of monomers. The taper can be either a graduation from conjugated diene rich chain to monovinylaromatic rich chain or a graduation from a monovinylaromatic rich chain to conjugated diene rich chain according to preference of the initiator-randomizer complex for one type monomer over the other. For example, when tetrahydrofuran is used as a randomizer, the diene enters into the chain faster than the monovinyl substituted aromatic; therefore, when both the monovinylaromatic monomer and the conjugated diene are present, the block tapers gradually from an essentially polybutadiene block to an essentially monovinyl substituted aromatic polymer block.

The weight ratio of monovinyl substituted aromatic monomer to conjugated diene monomer in each of the tapered blocks is 1:0.63 to 1:2, preferably 1:0.67 to 1:1.8, and more preferably 1:0.8 to 1:1.5. The weight ratios of monovinyl

substituted aromatic monomer to conjugated diene monomer in each of the tapered blocks in the same polymer chain do not have to be the same. See Example VII.

Generally each of the two tapered blocks made in steps (c) and (d) of the second embodiment of this invention can be of equal size; however, actual sizes of the two tapered blocks can vary within the same copolymer depending upon the amounts of monomers charged in each of the third and fourth monomer charges. See run 13 in Example VII.

Prior to coupling, all of the living polymer chains have monovinylaromatic terminal blocks on one end because of the initial monovinylaromatic charge (a) and charge (b) made with initiator; the living ends of the chains will have tapered blocks prior to coupling because of the charge containing both monovinylaromatic and conjugated diene monomers made next preceeding the coupling step.

In addition to the sequence of additions of the monomers and of the initiator, it is important to control the amount of each monomer and initiator addition at each stage or increment so that a suitable proportion of block sizes and proportion of polymodality is obtained.

Generally in a presently preferred first embodiment of this invention, if a polymer which is about 75 weight percent monovinylaromatic, based on total monomer weight, is used, 27 to 80 weight percent of the total monovinylaromatic monomer is charged in step (a), 5 to 21 weight percent of the total monovinylaromatic monomer is charged in step (b), and 7 to 60 weight percent of the total monovinylaromatic monomer is charged in step (c). Generally more preferably, 40 to 67 weight percent of the total monovinylaromatic monomer is charged in step (a), 8 to 19 weight percent of the total monovinylaromatic monomer is charged in step (b), and 27 to 40 weight percent of the total monovinylaromatic monomer is charged in step (c). Generally presently most preferably, 47 to 60 weight percent of the total monovinylaromatic monomer is charged in step (a), 11 to 16 weight percent of the total monovinylaromatic monomer is charged in step (b), and 29 to 37 weight percent of the total monovinylaromatic monomer is charged in step (c).

Generally in a presently preferred second embodiment of this invention, if a polymer which is about 75 weight percent monovinylaromatic, based on total monomer weight, is used, 27 to 80 weight percent of the total monovinylaromatic monomer is charged in step (a), 5 to 21 weight percent of the total monovinylaromatic monomer is charged in step (b), 3 to 19 weight percent of the total monovinylaromatic monomer is charged in step (c), and 13 to 36 weight percent of the total monovinylaromatic monomer is charged in step (d). Generally more preferably, 40 to 67 weight percent of the total monovinylaromatic monomer is charged in step (a), 8 to 19 weight percent of the total monovinylaromatic monomer is charged in step (b), 5 to 16 weight percent of the total monovinylaromatic monomer is charged in step (c), and 16 to 32 weight percent of the total monovinylaromatic monomer is charged in step (d). Generally presently most preferably, 47 to 60 weight percent of the total monovinylaromatic monomer is charged in step (a), 11 to 16 weight percent of the total monovinylaromatic monomer is charged in step (b), 8 to 13 weight percent of the total monovinylaromatic monomer is charged in step (c), and 20 to 25 weight percent of the total monovinylaromatic monomer is charged in step (d).

To either of the two embodiments of this invention it is feasible to stretch out over an interval of time the addition of one or more of the increments of initiator, thus spreading (increasing) further the polymodality of the resulting product upon coupling.

The polymerization process is carried out in a hydrocarbon diluent at any suitable temperature in a range of -10° to 150°C, preferably in the range of 0° to 110°C, at pressures sufficient to maintain the reaction mixture substantially in the liquid phase. Temperatures and pressures will peak during polymerization of each monomer charge and then decrease when essentially no free monomer is left to react. Appropriate hydrocarbon diluents include linear and cycloparaffins such as butane, pentane, hexane, octane, cyclohexane, cyclopentane, and mixtures thereof. Presently preferred is cyclohexane. Generally the choice of hydrocarbon or hydrocarbon mixture and the temperature is such that the resulting polymer is in solution.

Small amounts of polar compounds are used to improve the effectiveness of alkylmonoalkali metal initiators such as n-butyllithium; dissociation of the alkylmonoalkali metal initiators affects the rate of initiation and polymerization. The polar compounds also effect partial randomization of the vinylarene/conjugated diene so as to increase the random portion of the tapered block. The polar compounds are generally used in admixture with the hydrocarbon diluent.

The amounts of polar compounds used as randomizers and promoters of effectiveness of initiators in this invention will vary according to the reactivity and effectiveness of the particular randomizer used. For example, 1,2-dimethoxyethane, tetramethylethylenediamine and 1,2-dimethoxybenzene are much more efficient randomizers than most of the others listed above when used with the particular initiators and monomers used in the invention runs described below. However, tetrahydrofuran is often used because the reaction will go nearer to completion in a shorter time after the initial reaction in the monomer rich environment. Also, there are dramatic variations in the amounts of each of these most efficient randomizers which will be needed. For example, for polymerizations such as those shown in the examples of invention runs in Example I of this application, much less tetrahydrofuran than 1,2-dimethoxyethane would be needed.

The amounts of polar compounds used as randomizers will also vary according to the desired molecular structure of the portions of tapered blocks which result from conjugated diene addition. For example, when using tetrahydrofuran,

and it is desirable that the tapered blocks of the polymer have nearly equal amounts of 1,4 addition of butadiene and 1,2 addition of butadiene or only a little more 1,4 addition than 1,2 addition of butadiene, larger amounts of the tetrahydrofuran would be used than when more 1,4 addition is desired. When polymers with higher vinyl character resulting from 1,2 addition in excess of 1,4 addition are desired, then the useful amounts of tetrahydrofuran needed would be considerably larger. However, use of too much randomizer can result in excessive polymer-lithium termination during polymerization and/or poor stability of the polymer and/or undesired side reactions, depending upon choice of randomizer. Use of too little randomizer would result in inefficient initiator use, compositional variations and broader molecular weight distribution.

The initial monovinylaromatic charge is made with the randomizer present for the additional effect of causing the monovinylaromatic component resulting from each initiator charge to be of relatively narrow molecular weight distribution. In the two embodiments of this invention, by varying the amounts of initiator in each of the two charges having initiator, the differences in molecular weights of the monovinylaromatic components resulting from each of the two charges can be increased.

In each of the two embodiments of the invention, amounts of initiator employed are those which will produce resins with desirable melt flow which can be used in blends to make articles with a good balance of properties including minimal blueness, and good impact strength and/or ductility. Presently preferred when making invention polymers to be used in blends are amounts of initiator in each of the two initiator charges sufficient to obtain a block copolymer having a melt flow in the range from 2 to 50 g/10 minutes, more preferably from 4 to 30 g/10 minutes, and most preferably from 7 to 20 g/10 minutes, as determined by ASTM D1238-73, condition 200°C/5.0 kg. The amounts of initiator contemplated as useful in each of the two charges having initiator are shown in Tables 3 and 4.

Use of too small an amount of initiator would result in high molecular weight polymers. Conversely, use of too large an amount of initiator would result in polymers having short chain polymeric species and low molecular weight.

The weight ratio of the amounts of initiator in each of the charges having initiator can be represented as  $1:0.25-300$ .

$Li_1 : Li_2 :: 1 : 0.25-300$

wherein

$Li_1$  = initiator in the first charge

$Li_2$  = initiator in the second charge

More preferably for most applications, the amount of initiator in the second charge of initiator is from 0.3 to 10 times, based on weight, as much as the amount of initiator in the first initiator charge. Generally, presently most preferably, the amounts of initiators are selected such that the amount in the second charge is only slightly smaller, equal to, or larger than that employed in the first charge.

Varying the weight ratios of amounts of the initiator charges will result in variations of the proportionate amounts of species present in the copolymer. Other factors affecting the proportionate amounts of species present in the copolymer include presence of impurities and/or scavengers in the reactor, effectiveness of the polar randomizer as a promoter and choice of coupling agent(s).

The polymerization is carried out in substantial absence of oxygen and water, preferably under an inert gas atmosphere. Prior to the coupling step, the reaction mass contains a very high percentage of molecules (polymer chains) in which an alkali metal cation is positioned at one end of each polymer chain. Impurities in the feed such as water or alcohol reduce the amounts of monoalkali metal polymer in the reaction mass.

After essentially complete polymerization of the final charge added to the polymer, one or more suitable difunctional or polyfunctional coupling agents is added. As used here, the term "coupling" means the bringing together and joining, by means of one or more central coupling atoms or coupling moieties, two or more of the living monoalkali metal-terminated polymer chains. A wide variety of compounds for such purposes can be employed.

Any effective amount of the coupling agent can be employed. While the amount is not believed to be particularly critical, a stoichiometric amount relative to the active polymer-alkali metal tends to promote maximum coupling as a generality. Presently preferred is an amount of coupling agent slightly greater than stoichiometric relative to the active polymer-alkali metal. However, less than stoichiometric amounts can be used for higher degrees of coupling where desired for particular products of broadened molecular weight distribution.

Typically, in each of the embodiments of this invention, the total amount of coupling agent is in the range of 0.005 to 10 phm (parts per 100 parts of total monomers employed in the polymerization). Preferred when most combinations of monomers and coupling agents are used to practice this invention is 0.2 to 0.6 phm of coupling agent, depending on amounts of initiator used. Presently most preferred is 0.3 to 0.5 phm, depending upon amounts of initiator used. Use of an amount of a reasonably highly efficient coupling agent within these ranges provides polymers with a moderately broad molecular weight distribution that has proven useful to custom molders. Use of an insufficient amount of coupling agent will result in less complete coupling of the living polymer chains or, depending upon choice of coupling agent, more branching; use of an excessive amount of coupling agent will also result in more uncoupled chains.

At the conclusion of the coupling process, the coupled polymer may still contain bound polymeric alkali metal alkoxides depending on the type of coupling agent employed. The system is treated with an active compound such as water, alcohol, phenols, carbon dioxide or linear saturated aliphatic mono- and dicarboxylic acids to remove any remaining alkali metal from the copolymer chain.

5 While the polymer product is still in solution, stabilization agents can be added. Additional stabilizers could be added during finishing before pelletizing. This treatment will provide oxidative stability for the polymer during processing and handling and subsequent long term use by the customer.

Commonly used stabilization processes can use a combination of compounds which include, but are not limited to, a hindered phenol and an organophosphite, particular examples of which are octadecyl 3-(3',5'-di-*t*-butyl-4'-hydrox-  
10 yphenyl) propionate and tris-nonylphenylphosphite.

After stabilization, the hydrocarbon diluent is then flashed from the polymer solution to increase the solids content. The polymer cement, i.e., the polymer in the polymerization solvent, usually contains 10 to 40, more usually 20 to 35, weight percent solids, the balance solvent. Preferably, but not necessarily, the polymer cement is flashed to remove by evaporation a portion of the solvent so as to reduce the solvent content to a concentration of 0 to 50, more usually  
15 0 to 10, weight percent (corresponding to a solids content of 100 to 50, more usually 100 to 90, weight percent).

Flashing of the polymer cement may be followed by desolventizing extrusion with vacuum in commercial production or by other vacuuming processes to achieve consistent solvent content of less than 0.3 weight percent.

The resinous copolymeric products can be, and normally are, compounded with anti-oxidants, anti-blocking agents, release agents and other additives, as known in the compounding arts.

20 Typical charging sequences and useful ranges of amounts of the charges for each of the two embodiments of this invention are given in Tables 3 and 4.



TABLE 3

Ranges of Amounts of Components in a Typical Invention Charging Sequence<sup>a</sup>  
(First Embodiment)

Step	Component <sup>b</sup>	Broad Range <sup>c</sup>	Preferred Range <sup>c</sup>	More Preferred Range <sup>c</sup>
(a)	randomizer <sup>d</sup> initiator and monovinylarene monomer	0.001-3.0 0.005-1.5	0.005-1.5 0.01 -0.1	0.015-0.3 0.01 -0.05
		20-60	30-50	35-45
(b)	initiator and monovinylarene monomer	0.005-1.5	0.01-1.0	0.02-0.1
		4-16	6-14	8-12
(c)	monovinylarene monomer and conjugated diene monomer	5-45	20-30	22-28
		5-45	20-30	22-28
(e) *	coupling agent	0.005-10	0.2-0.6	0.3-0.5

<sup>a</sup>Charges of monomer and initiator are made in the presence of an amount of diluent or solvent sufficient to prevent excessive heat of reaction.

<sup>b</sup>Monomers within each charge having more than one monomer can be added simultaneously or as a mixture, slowly or quickly. Randomizer in charge (a) may be already present in the diluent or added as a separate charge.

<sup>c</sup>Ranges of amounts are given in parts by weight per 100 parts by weight of total monomers (phm).

<sup>d</sup>To achieve high vinyl content, up to 3 phm randomizer may be used.

\* step (d) is optional according to claim 1 and not conducted according to the first embodiment

TABLE 4

Ranges of Amounts of Components in a Typical Invention Charging Sequence<sup>a</sup>  
(Second Embodiment)

Step	Component <sup>b</sup>	Broad Range <sup>c</sup>	Preferred Range <sup>c</sup>	More Preferred Range <sup>c</sup>
(a)	randomizer <sup>d</sup> initiator and monovinylarene monomer	0.001-3 0.005-1.50  20-60	0.005-1.85 0.01 -0.1  30-50	0.015-0.3 0.04 -0.08  35-45
(b)	initiator and monovinylarene monomer	0.005-1.5	0.01 -1.0	0.03 -0.07
(c)	monovinylarene monomer and conjugated diene monomer	4-16  2-14 2-13	6-14  4-12 3-11	8-12  6-10 5-9
(d)	monovinylarene monomer and conjugated diene monomer	10-27  8-28	12-24  13-23	15-19  16-20
(e)	coupling agent	0.005-10	0.2-0.6	0.3-0.5

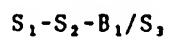
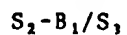
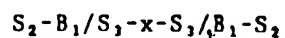
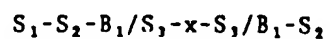
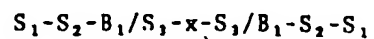
<sup>a</sup>Charges of monomer and initiator are made in the presence of an amount of diluent or solvent sufficient to prevent excessive heat of reaction.

<sup>b</sup>Monomers within each charge having more than one monomer can be added simultaneously or as a mixture, slowly or quickly. Randomizer in charge (a) may be already present in the diluent or added as a separate charge.

<sup>c</sup>Ranges of amounts are given in parts by weight per 100 parts by weight of total monomers (phm).

<sup>d</sup>To achieve high vinyl content, up to 3 phm randomizer may be used.

After coupling at least the coupled and uncoupled polymeric species shown in Tables 5 and 6 are present in the polymodal polymers of the first and second embodiments, respectively, of this invention.

TABLE 5Polymeric Species Included in First Embodiment

S = monovinylaromatic block

B = conjugated diene block

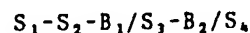
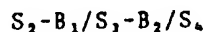
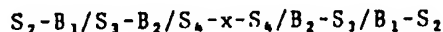
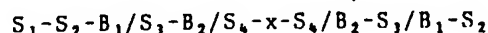
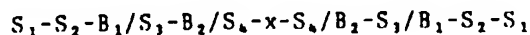
B/S = tapered block

x = coupling moiety or coupling site

subscripts = indications of the charges which were the source of the polymer blocks.

TABLE 6

## Polymeric Species Included in Second Embodiment



S = monovinylaromatic block

B = conjugated diene block

B/S = tapered block

x = coupling moiety or coupling site

subscripts = indications of the charges which were the source of the polymer blocks.

In each of the embodiments depending upon choice and amount of coupling agent or agents and whether coupling agents are charged as a mixture or incrementally, there can be present other polymeric species with varying degrees of branching.

#### Blends

The resinous polymodal copolymer products of this invention can be blended with other polymers such as acrylonitrile-butadiene-styrene copolymers (ABS), styrene-acrylonitrile copolymers (SAN), and other styrene copolymers. When less transparency is desired or transparency is not necessary, the invention copolymers can be blended with polyolefins and/or olefin copolymers.

Blends of the invention polymers with polymers of styrene are particularly useful applications for the improved tapered block copolymers of this invention. Articles made from these blends have surprisingly low blueness and advantageous impact strength and/or ductility. For example, articles made from styrene homopolymer and either of the two embodiments of this invention typically have Hunter negative b blueness values of less than 20, most often less than 18. Thus, less of the invention copolymer is needed to achieve high impact strength, while retaining reasonably low blueness, in articles made from blends with styrene homopolymers than would be needed of other comparative polymodal monovinylaromatic-conjugated diene block copolymers.

The presently preferred polymers of styrene employed in the blends of this invention are usually (a) homopolymers of styrene; or (b) copolymers of styrene as a major component with a minor amount of any other copolymerizable monovinyl aromatic compound other than styrene, such as alpha-methylstyrene, vinyltoluene or para-tertiary-butyl styrene. A minor amount of other monomers such as methyl acrylate, methyl methacrylate and acrylonitrile can be copolymerized with the styrene.

The invention copolymers can be blended with styrene resins made in bulk polymerization. These resins are commonly prepared by heating styrene and any comonomer at temperatures in the range of 100° to 200°C with application of pressure, if necessary, to combine the monomers. The polymerization can also be carried out at lower temperatures by the addition of free-radical generating peroxidic catalysts such as benzoyl peroxide, acetyl peroxide and di-t-butyl peroxide. Alternatively, the polymerization can be carried out in suspension to yield a dry powder or in emulsion, usually resulting in a latex of polystyrene which can be coagulated to yield the solid powdery polystyrene. The polymerization

can also be carried out in solution with precipitation of the product, if desired. Solvent can be removed by standard techniques such as steamstripping or solvent evaporation.

High impact polystyrene (HIPS) also can be successfully employed in blends with the invention copolymers. Suitable high impact polystyrenes can be prepared by polymerizing styrene in the presence of an elastomer, typically polybutadiene rubber. In these resins the polystyrene forms the bulk of the resin throughout which the grafted rubber polystyrene particles are dispersed.

The blends of this invention can be prepared by any suitable means including blending, tumbling and extrusion. Examples of these methods include, but are not limited to, dry mixing in the form of a powder or pellets, wet mixing in the form of a solution or slurry, and melt extrusion compounding.

The polymers and any other ingredients or additives may be mechanically blended together in the desired proportions with the aid of any suitable mixing device conveniently used for mixing rubbers or plastics, such as, for example, a differential roll mill, a Banbury mixer, or an extruder.

In these types of blending methods the polymers and any other components and additives used can be in any form, such as, for example, fluff, powder, granulate, pellet, solution, slurry, and/or emulsion. Any additive can be combined with the polymers according to any method known in the art. Examples of incorporation methods include, but are not limited to, dry mixing in the form of a powder and wet mixing in the form of a solution or slurry.

Melt extrusion compounding can be carried out using any suitable method such as in single screw or twin screw extruders or other melt extruders at temperatures above the melting point or glass transition temperature of the polymers.

The presently preferred method comprises blending the polymers in powder or granulate form and extruding the blend in sheet form to feed a thermoformer or direct feed to an injection or blow molder.

In order to facilitate thorough mixing of the polymers and to develop the desired combination of physical properties, pellets are generally metered by loss-in-weight feeders or by screw feeders at a temperature low enough to avoid softening the pellets. The metered pellets are dropped into an extruder which melts and blends the components to provide a homogenous melt.

Alternatively, solution blending methods known in the art may be used.

The ranges of amounts of polymers useful in blends varies according to the properties and economics required. For example, when an invention copolymer is blended with a polymer of styrene, practical ranges include using amounts such as 10 to 70 weight percent polymer of styrene, more usually 20 to 65 weight percent polymer of styrene, and most preferably 30 to 60 weight percent polymer of styrene with the balance being one or more of the polymodal resinous copolymer products of this invention. For a more particular example, when the invention copolymers are blended with general purpose polystyrene, broad ranges include using amounts such as 10 to 70 weight percent polystyrene, more usually 20 to 65 weight percent polystyrene, and most preferably 30 to 60 weight percent polystyrene with the balance being one or more of the polymodal resinous copolymer products of this invention. Generally, use of too much of the invention copolymers in a blend would result in good properties but loss of economic advantage. Use of too little of the invention copolymers in a blend would result in loss of impact resistance. These blends can be economical ways of gaining the desirable attributes of both polystyrene and the invention polymers while maintaining low blueness in articles made from the blends.

The compositions of this invention or blends thereof can be extruded, thermoformed, injection molded, blow molded, or made into films or sheets. Articles made from the compositions of this invention are transparent with low blueness, have good impact strength and have other physical properties within acceptable ranges for such applications as drinking cups, lids, bottles, other food containers, medical drainage units, shrink wrap and over-wrap. Articles made from blends of the copolymers of this invention can economically provide advantageous properties for similar applications.

#### Test Procedures

The blend blueness values were determined on a Hunter Lab colormeter Model D 25 using the Hunter Lab procedure. Blueness values are expressed as -b, where larger absolute numbers indicate more blueness.

Other properties were tested using ASTM procedures as shown in Table 7.

TABLE 7

Test Procedures Used	
Property	ASTM Method
Flow rate, g/10 min Condition 200°C/5.0 kg	D 1238-88
Haze, %	D 1003-61 (1990)
Transmittance, %	D 1003-61 (1990)

TABLE 7 (continued)

Test Procedures Used	
Property	ASTM Method
Shore D hardness	D 2240-91
Tensile strength at yield and break, MPa	D 638-91 at 50 mm/min Type I test specimens
Elongation at yield and break, %	D 638-91 at 50 mm/min
Flexural modulus, MPa	D 790-86
Izod impact strength, notched, J/m	D 256-88
Vicat softening point, °C	D 1525-91
Total energy dart drop, J	D 4272-85

**Examples**

The following examples will describe in more detail the experimental process used and the polymodal tapered block copolymers with vinylarene terminal blocks obtained as a result of the process and blends made with the invention copolymers. These examples should be taken as illustrative and not restrictive.

Styrene and butadiene were chosen as monomers to exemplify the invention, and randomizer, initiator, coupling agent and diluent appropriate for these monomers were used. Quantities of reagents are usually expressed in parts per hundred monomer (phm) based on the total weight of monovinylarene and conjugated diene employed.

**Example I**

This example describes four invention polymerization runs (1, 2, 3, 4) that were carried out to produce resinous polymodal, coupled, tapered block styrene-butadiene copolymers with resinous terminal blocks. These runs exemplify the first embodiment of the invention. Styrene (from Stirling Chemical) and 1,3-butadiene (from Texas E1 Paso) were dried by passage over activated alumina (Kaiser A-201), and then copolymerized and coupled in a 4-stage process using n-butyllithium initiator (from Lithium Corporation of America).

Polymerization runs were carried out under nitrogen in a stirred, jacketed, stainless steel 7.6-liter reactor employing essentially anhydrous reactants and conditions. The anhydrous mixtures were stirred continuously during the polymerization process. The cyclohexane diluent, which contained 0.04 phm tetrahydrofuran (THF) in each polymerization in this example, was preheated to about 50°C before monomers were charged to the reactor. The n-butyllithium was charged as a 2 weight percent solution in cyclohexane. In the polymerization step in which both butadiene and styrene were charged, they were charged simultaneously as a mixture.

In the coupling step, the Vikoflex® 7170 coupling agent used was an epoxidized vegetable oil commercially available from Viking Chemical Company. In the terminating step, carbon dioxide from a pressurized container was admitted to provide about 0.4 phm carbon dioxide to the reactor. Water was also added in an amount slightly in stoichiometric excess of the initiator to separate the lithium residues from the polymer chains.

The antioxidant mixture added in the stabilizing step contained a hindered phenol [octadecyl 3-(3',5'-di-*t*-butyl-4'-hydroxyphenyl) propionate, commercially available as Irganox® 1076 from Ciba-Geigy] and an organic phosphite (tris-nonylphenyl phosphite, available as TNPP from GE Specialty Chemicals). Each stabilizer was dissolved separately in cyclohexane and mixed together. Enough of the mixture was added to the reactor to provide 0.25 phm hindered phenol and 1 phm organic phosphite. In Runs 1 and 2 a microcrystalline wax (BE Square® 195) was also added as an anti-blocking agent.

After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 10-20 phm cyclohexane solvent and cleared with a nitrogen sparge.

Following the stabilization step, each copolymer solution was flashed at 178-180°C to remove a portion of the diluent. Substantially all of the remaining diluent was removed in a vacuum oven by drying at 90°C for one hour. The resulting polymer was chopped in a granulator into crumb size and then dried for an additional hour in a vacuum oven.

In each of the four runs (1, 2, 3, and 4), 1500 g total monomers (butadiene and styrene) were used. About 76 weight percent of the total amount of cyclohexane diluent (3130 g) was charged initially. The remaining cyclohexane diluent was added during the run as a diluent or flush for the various reactants added in subsequent steps. In these four polymerizations, the weight ratio of total monomers charged was 75/25 styrene/butadiene.

The charges and the results of the runs are summarized in Table 8. Tapered butadiene/styrene blocks were formed in step 3 by charging both butadiene and styrene monomers. The charging sequence used was i, S<sub>1</sub>, i, S<sub>2</sub>, B<sub>1</sub>/S<sub>3</sub>, coupling agent. The monomer ratios corresponding to the S<sub>1</sub>, S<sub>2</sub>, B<sub>1</sub>/S<sub>3</sub> sequence were 40, 10, 25/25. Weight ratios

of amounts of initiator used in the first two steps of each of the four runs was kept constant at 1:1, but the amounts were decreased from 0.08 phm in run 1 to 0.055 phm in run 4 to provide a series of decreasing melt flow resins.

The devolatilized copolymers from runs 1, 2, 3, and 4 were designated invention copolymers 1, 2, 3 and 4, and had melt flows of 39.1 g/10 min, 19.0 g/10 min, 9.0 g/10 min and 4.8 g/10 min, respectively. The differences in melt flows were attributable to differences in amounts of initiator used in each of the four runs.

TABLE 8

Invention Runs - First Embodiment				
Components <sup>a</sup>	Run 1	Run 2	Run 3	Run 4
Step 1				
Cyclohexane, phm	145	145	145	145
Tetrahydrofuran, phm	0.04	0.04	0.04	0.04
n-Butyllithium initiator, phm	0.08	0.07	0.06	0.055
Styrene, phm	40	40	40	40
Polymerization Time, min	12	12	12	12
Peak Polymerization Temperature, °C	82	81	85	85
Step 2				
n-Butyllithium initiator, phm	0.08	0.07	0.06	0.055
Styrene, phm	10	10	10	10
Polymerization Time, min	12	12	12	12
Peak Polymerization Temperature, °C	67	70	71	70
Step 3				
Butadiene, phm	25	25	25	25
Styrene, phm	25	25	25	25
Polymerization Time, min	16	16	16	16
Peak Polymerization Temperature, °C	122	122	122	116
Step 4 (Coupling)				
Vikoflex 7170, phm	0.4	0.4	0.4	0.4
Time, min	16	16	16	16
Temperature, °C	94	93	90	93
Step 5 (Terminating)				
Water, phm	0.2	0.2	0.2	0.2
Carbon Dioxide, phm	0.4	0.4	0.4	0.4
Time, min	25	25	25	25
Temperature, °C	80	93	87	81
Step 6 (Stabilizing)				
Stabilizer Mixture, phm	1.25	1.25	1.25	1.25
BE Square wax	0.15	0.15	0.15	0.15
Time, min	5	5	5	5
Temperature, °C	n.a.	n.a.	84	n.a.

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

TABLE 8 (continued)

Invention Runs - First Embodiment				
Components <sup>a</sup>	Run 1	Run 2	Run 3	Run 4
Recovered Resin				
Melt Flow, g/10 min	39.1	19.0	9.0	4.8
Mw/Mn, thousands	105/63	108/53	128/65	143/91
Heterogeneity Index	1.67	2.05	1.97	1.57

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

**Example II**

Three more polymerization runs were carried out according to the first embodiment of the present invention to demonstrate the effects of varying the weight ratio of amounts of initiator in each of the two initiator charges. The charges and results of the runs are shown in Table 9. Again, the weight ratio of styrene to butadiene charged was 75 to 25. Samples were made with i, S<sub>1</sub>, i, S<sub>2</sub>, B<sub>1</sub>/S<sub>3</sub> addition sequence followed by coupling; monomer ratios of 40, 10, 25/25 were used.

In each of the three runs of this example, 0.03 phm initiator was charged in the first step. The amount of initiator charged in the second step was varied from 0.08 phm (run 5) to 0.95 phm (run 7) for a range of ratios of amount of initiator in first step to amount in second step from 1:2.7 (run 5) to 1:3.2 (run 7).

Copolymers from runs 5, 6 and 7 were devolatilized to form invention copolymers 5, 6 and 7, which had melt flows of 5.0 g/10 min, 5.4 g/10 min, and 7.1 g/10 min, respectively.

TABLE 9

Invention Runs - First Embodiment			
Components <sup>a</sup>	Run 5	Run 6	Run 7
Step 1			
Cyclohexane, phm	145	145	145
Tetrahydrofuran	0.04	0.04	0.04
n-Butyllithium initiator, phm	0.03	0.03	0.03
Styrene, phm	40	40	40
Polymerization Time, min	12	12	12
Peak Polymerization Temperature, °C	77	79	76
Step 2			
n-Butyllithium initiator, phm	0.08	0.085	0.095
Styrene, phm	10	10	10
Polymerization Time, min	12	12	12
Peak Polymerization Temperature, °C	64	71	65
Step 3			
Butadiene, phm	25	25	25
Styrene, phm	25	25	25
Polymerization Time, min	16	16	16
Peak Polymerization			

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.



TABLE 9 (continued)

Invention Runs - First Embodiment			
Components <sup>a</sup>	Run 5	Run 6	Run 7
Step 3			
Temperature, °C	117	121	111
Step 4 (Coupling)			
Vikoflex 7170, phm	0.4	0.4	0.4
Time, min	16	16	16
Temperature, °C	88	87	89
Step 5 (Terminating)			
Water, phm	0.2	0.2	0.2
Carbon Dioxide, phm	0.4	0.4	0.4
Time, min	25	25	25
Temperature, °C	82	82	84
Step 6 (Stabilizing)			
Stabilizer Mixture, phm	1.25	1.25	1.25
BE Square wax	0.15	0.15	0.15
Time, min	5	5	5
Recovered Resin			
Melt Flow, g/10 min	5.0	5.4	7.1
Mw/Mn, thousands	181/114	177/111	138/93
Heterogeneity Index	1.6	1.6	1.69

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

### Example III

To demonstrate the second embodiment of this invention, three more polymerization runs were carried out according to the procedures described in Example I, with the exception that the sequences and amounts of charges were as shown in Table 10. Tapered butadiene/styrene blocks were formed in the third and fourth steps by charging a mixture of butadiene and styrene monomers. The monomer addition sequence was i, S<sub>1</sub>, i, S<sub>2</sub>, B<sub>1</sub>/S<sub>3</sub>, B<sub>2</sub>/S<sub>4</sub> followed by coupling; the monomer weight ratios were 40, 10, 12.5/12.5, 12.5/12.5, respectively. The polymers were 75 percent styrene and 25 percent butadiene.

In each of the three runs of this example the weight ratio of amount of initiator in the first step to the amount in the second step was kept constant at 1:1 with the absolute amount varied from 0.06 phm (run 8) to 0.055 phm (run 10).

The copolymers produced in the three runs were designated invention copolymers 8, 9, and 10, and had melt flows of 14.3, 6.40 and 10.8 respectively. It is believed that lower melt flows are attributable to lower amounts of initiator. The polymerizations of invention runs 8, 9 and 10 are shown in Table 10.

TABLE 10

Invention Runs - Second Embodiment			
Components <sup>a</sup>	Run 8	Run 9	Run 10
Step 1			
Cyclohexane, phm	145	145	145
Tetrahydrofuran, phm	0.04	0.04	0.04

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

TABLE 10 (continued)

Invention Runs - Second Embodiment				
Components <sup>a</sup>		Run 8	Run 9	Run 10
5	Step 1			
	n-Butyllithium initiator, phm	0.06	0.05	0.055
	Styrene, phm	40	40	40
	Polymerization Time, min	12	12	12
10	Peak Polymerization Temperature, °C	85	82	84
	Step 2			
15	n-Butyllithium initiator, phm	0.06	0.05	0.055
	Styrene, phm	10	10	10
	Polymerization Time, min	12	12	12
20	Peak Polymerization Temperature, °C	69	69	70
	Step 3			
25	Butadiene, phm	12.5	12.5	12.5
	Styrene, phm	12.5	12.5	12.5
	Polymerization Time, min	16	16	16
	Peak Polymerization Temperature, °C	84	83	84
30	Step 4			
	Butadiene, phm	12.5	12.5	12.5
	Styrene, phm	12.5	12.5	12.5
	Polymerization Time, min	16	16	16
35	Peak Polymerization Temperature, °C	96	102	102
	Step 5 (coupling)			
40	Vikoflex 7170, phm	0.4	0.4	0.4
	Time, min	16	16	16
	Temperature, °C	82	89	89
	Step 6 (Terminating)			
45	Water, phm	0.2	0.2	0.2
	Carbon Dioxide, phm	0.4	0.4	0.4
	Time, min	25	25	25
	Temperature, °C	80	82	82
50	Step 7 (Stabilizing)			
	Stabilizer Mixture, phm	1.25	1.25	1.25
	Antiblocking agent, phm	0.15	0.15	0.15
55	Time, min	5	5	5

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

TABLE 10 (continued)

Invention Runs - Second Embodiment			
Components <sup>a</sup>	Run 8	Run 9	Run 10
Step 7 (Stabilizing)			
Temperature, °C	80	n.a.	81
Recovered Resin			
Melt Flow, g/10 min	14.3	6.4	10.8
Mw/Mn, thousands	154/104	173/115	147/91
Heterogeneity Index	1.48	1.50	1.62

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

#### Example IV

Two comparative polymers were used for comparisons of physical properties of blends made with polystyrene. Comparative polymer 11 was a resinous styrene-butadiene copolymer with a melt flow of 8.4 g/10 min. Polymer 11 was polymodal from multiple initiator and monomer charges (S, i, i, S, B, i, S, B) and coupled with an epoxidized vegetable oil coupling agent. Polymer 11 contained nominally 75 wt % styrene and 25 wt % butadiene with no styrene/butadiene tapered blocks. Polymer 11 forms 50:50 by weight blends with polystyrene that have high blueness and modest physical properties.

Comparative polymer 12 was a styrene-butadiene copolymer with no tapered block segments. Polymer 12 contained nominally 75 wt % styrene and 25 wt % butadiene and had a melt flow of 8.8 g/10 min. It had a polymodal molecular weight from multiple initiator and monomer charges (S, i, i, S, B) and was coupled with an epoxidized vegetable oil coupling agent. Polymer 12 formed 50:50 blends by weight with polystyrene that have low blueness and modest impact properties.

#### Example V

A series of blends of the invention polymers from Examples I and II was prepared with Novacor® 555 general purpose polystyrene to demonstrate the properties of blends of copolymers containing a single tapered block (invention embodiment 1). Polymer 5 was not included in the blend series since it had a melt flow essentially the same as that of polymer 4. Polymers 11 and 12 were also blended with polystyrene for comparison.

The blends were 50:50 by weight and were prepared by solution blending in cyclohexane in a reactor at 100°C with stirring for one hour. Each solution was flashed to remove the solvent and the polymer was dried, chopped in a granulator, and dried again. The dry polymer was processed on a roll mill and chopped again before injection molding on an Arburg model 90 injection molder with zones 1, 2, 3, and nozzle at 210°C, 210°C, 220°C, and 210°C, respectively, to make specimens to test for properties.

For convenient reference, blends are designated as blends with the corresponding polymer number and a prime. For example, a blend prepared from polymer 1 and polystyrene is blend 1' and a blend prepared from polymer 2 and polystyrene is designated blend 2'.

The blend test results are shown in Table 11. Test specimens made from the first four blends had less blueness than test specimens made from the comparative polymers. Specimens made from blends 6' and 7' had essentially the same blueness as specimens made from blend 12'.

Test specimens made from blends 1', 2', 3', 4', and 6' had Izod impact values higher than those made from comparative blends 11' and 12'. Test specimens made from blend 7' had a lower Izod impact value than those made from the other blends. It is possible that the high level of initiator in step two of the polymerization of polymer 7 and the resulting higher melt flow of the final resin prevented improvement in the Izod impact value of Arburg test specimens made from blends polymer 7.

Test specimens made from blends 1', 2', and 3' had higher haze values and specimens made from blends 1' and 2' had higher melt flows than those made from the other blends.

TABLE 11

Physical Properties of Test Specimens Made from Polymer Blends with Polystyrene<sup>a</sup>

Property	Invention Blend 1'	Invention Blend 2'	Invention Blend 3'	Invention Blend 4'	Invention Blend 6'	Invention Blend 7'	Comparative Blend 11'	Comparative Blend 12'
Styrene-butadiene copolymer	1	2	3	4	6	7	11	12
Blend melt flow, g/10 min	35.5	30.2	19.0	11.5	10.7	12.8	16.8	17.5
Haze, %	12.3	8.1	7.1	5.3	3.4	4.2	4.3	2.7
Hunter blueness, b	-15.5	-16.2	-15.7	-16.3	-16.9	-17.1	-20.3	-16.8
Shore D Hardness	76	77	78	77	77	77	77	79
Notched Izod impact, J/m	14.5	14.6	18.1	20.3	17.1	12.3	12.4	13.9
Vicat, °C	83.4	82.9	86.8	89.8	92.1	92.0	85.3	89.3
Tensile strength								
yield, MPa	39.4	40.2	41.4	43.7	44.5	42.5	37.8	42.6
break, MPa	24.8	27.5	29.8	33.6	33.7	21.4	28.0	33.4
Elongation								
yield, %	5.2	5.3	5.3	6.6	6.2	6.2	4.8	4.8
break, %	17.7	18.2	24.9	15.0	14.7	13.7	32.0	15.77
Flexural modulus, MPa	1754	1836	1862	2107	2091	2016	1979	2124

<sup>a</sup>50:50 Styrene-butadiene copolymer:polystyrene

**Example VI**

A series of blends of double tapered block invention polymers (second invention embodiment) with polystyrene was prepared to demonstrate the properties of articles made from the blends. Copolymers 8, 9, and 10 from Example III were blended as described in Example V in 50:50 by weight mixtures with general purpose polystyrene to form blends 8', 9', and 10'. Test specimens were injection molded from the blends as described in Example V.

The results (Table 12) show that articles made from blends of invention polymers 8, 9, and 10 had better Izod impact values than those made from comparative blends 11' and 12' (shown in Table 11). Articles made from blends of the invention polymers also have less blueness than those made from comparative blend 11'.

TABLE 12

Physical Properties of Polymer Blends With Polystyrene <sup>a</sup>				
Property	Invention Blend 8'	Invention Blend 9'	Invention Blend 10'	Comparative Blend 11'
Styrene-butadiene copolymer	8	9	10	11
Blend melt flow, g/10 min	14.8	12.0	14.8	16.8
Haze, %	4.1	4.0	5.1	4.3
Hunter blueness, b	-17.9	-18.5	-17.9	-20.3
Shore D hardness	76	77	76	77
Notched Izod				
impact, J/m	14.9	17.1	19.2	12.3
Vicat softening, °C	89.3	90.5	90.5	85.3
Tensile Strength				
yield MPa	43.2	43.9	42.5	37.8
break MPa	28.6	28.7	28.7	28.0
Elongation				
yield, %	6.4	6.5	6.5	4.8
break, %	17.9	16.7	13.6	32.0
Flexural				
modulus, MPa	1988	1949	1986	1979

<sup>a</sup>50:50 Styrene-butadiene copolymer:polystyrene

**Example VII**

Three more invention copolymers were prepared on a larger scale to demonstrate further the effect of two butadiene/styrene tapered blocks in improving impact properties of articles made from blends of the invention copolymers with polystyrene. These polymerizations varied the size of the two tapered blocks and the weight ratios of initiator charges. The polymerizations were carried out in a 380 L reactor.

The polymer polymerization recipes are shown in Table 13. Each polymer was prepared with the sequence i,S,i,S,B/S,B/S. Polymer 13 had monomer charge weight ratios of 40, 10, 6.9/7.8, 18.1/17.2. Polymer 14 had monomer charge weight ratios of 40, 10, 5.9/10, 19.1/14.4. Polymer 15 had monomer charge ratios of 40, 10, 5.9/10, 19.1/14.4. Monomer charge ratios for polymers 14 and 15 do not add up to 100 phm because of charge irregularities; phm indicated above was that actually charged. The initiator weight ratios for steps 1:2 in polymerizations 13, 14, and 15 were 1:0.79, 1:0.92, and 1:0.89, respectively.

Flow rates for the three polymers were 7.0, 11.5, and 13.7 g/10 min.

TABLE 13

Invention Runs - Second Embodiment				
	Run 13	Run 14	Run 15	
5				Step 1
	168.9	168.9	168.9	Cyclohexane, phm
	.04	.04	0.04	Tetrahydrofuran, phm
10				n-Butyllithium
	0.061	0.061	0.065	initiator, phm
	40	40	40	Styrene, phm
	1.1	1.1	1.1	Cyclohexane, phm
				Peak Polymerization
15	83	83	83	Temperature, °C
	214 (31)	221 (32)	207 (30)	Peak Polymerization Pressure, kPa (psi)
				Step 2
20				n-Butyllithium
	0.048	0.056	0.058	initiator, phm
	10	10	10	Styrene, phm
	1.1	1.1	1.1	Cyclohexane, phm
25				Peak Polymerization
	81	82	82	Temperature, °C
	242 (35)	255 (37)	235 (34)	Peak Polymerization Pressure, kPa (psi)
				Step 3
30				Butadiene, phm
	6.9	5.9	5.9	Styrene, phm
	7.8	10	10	Cyclohexane, phm
	1.1	1.1	1.1	Peak Polymerization
35	110	109	109	Temperature, °C
	394 (57)	400 (58)	380 (55)	Peak Polymerization Pressure, KPa (psi)
				Step 4
40				Butadiene, phm
	18.1	19.1	19.1	Styrene, phm
	17.2	14.4	14.4	Cyclohexane, phm
	1.1	1.1	1.1	Peak Polymerization Temperature, °C
45	95	98	97	Peak Polymerization Pressure, kPa (psi)
	359 (52)	373 (54)	345 (50)	
				Polymer Analysis (prior to coupling)
				Mw/Mn, thousands <sup>a</sup>
				(GPC area composition) <sup>b</sup>
50	104/99 (64)	100/95 (63)	95/90 (64)	Peak 1
	50/46 (36)	47/44 (37)	46/43 (36)	Peak 2
				Step 5 (Coupling)
				Vikoflex 7170, phm
55	0.4	0.4	0.4	Cyclohexane, phm
	0.55	0.55	0.55	

<sup>a</sup>Polystyrene equivalent Mw/Mn. Mw/Mn of polystyrene with the same hydrodynamic volume as the polymer peak.<sup>b</sup>Response of a dielectric detector in an Applied Automation process control GPC.

TABLE 13 (continued)

Invention Runs - Second Embodiment			
	Run 13	Run 14	Run 15
Step 5 (Coupling)			
Temperature, °C	88	86	88
Pressure, kPa (psi)	345 (50)	359 (52)	331 (48)
Step 6 (Terminating)			
Water, phm	0.2	0.2	0.2
Carbon Dioxide, phm	0.1	0.1	0.1
Temperature, °C	87	87	87
Pressure, kPa (psi)	345(50)	345(50)	345(50)
Step 7 (Stabilizing)			
Stabilizer Mixture, phm	1.25	1.25	1.25
Antiblocking agent, phm	0.3	0.3	0.3
Recovered Resin			
Melt Flow, g/10 min	7.0	11.5	13.7

**Example VIII**

Two comparative polymers (16 and 17) were prepared for comparison with the invention polymers from Example VII. Polymer 16 was a coupled polymodal styrene/butadiene copolymer containing 75 wt % styrene and 25 wt % butadiene. Polymer 16 was prepared with the sequence i,S,i,S,B/S,i,S,B using monomer charge weight ratios of 37, 19, 7.5/5, 14, 17.5, respectively.

Polymer 17 was a coupled polymodal styrene/butadiene copolymer containing 70 wt % styrene and 30 wt % butadiene. Polymer 17 was prepared with the sequence i,S,i,S,B,i,S,B using monomer charge weight ratios of 37, 19, 9, 14, 21, respectively.

These comparative polymers were prepared in a 380 L reactor according to the procedure in Table 14.

Polymer 16 yielded blends with polystyrene which were made into test specimens that had low blueness and modest physical properties. Polymer 17 formed blends with polystyrene which were made into test specimens that had high blueness and good physical properties.

The flow rates for polymers 16 and 17 were 8.2 and 8.6 g/10 min, respectively.

TABLE 14

Invention Runs - Second Embodiment		
Components <sup>a</sup>		
Step 1		
Cyclohexane, phm	168	168
Tetrahydrofuran, phm	0.02	0.02
n-Butyllithium		
initiator, phm	0.034	0.031
Styrene, phm	37	37
Peak Polymerization		
Temperature, °C	88	87
Peak Polymerization Pressure, kPa (psi)	255 (37)	248 (36)
Step 2		
n-Butyllithium	0.048	0.053

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.

TABLE 14 (continued)

Invention Runs - Second Embodiment			
Components <sup>a</sup>		Run 16	Run 17
5	Step 2		
	initiator, phm		
	Styrene, phm	19	19
	Peak Polymerization		
10	Temperature, °C	86	85
	Peak Polymerization Pressure, kPa (psi)	228 (33)	221 (32)
	Step 3		
15	Butadiene, phm	7.5	9
	Styrene, phm	5	0
	Peak Polymerization		
	Temperature, °C	82	83
20	Peak Polymerization Pressure, kPa (psi)	255 (37)	249 (36)
	Step 4		
	n-Butyllithium initiator, phm	0.085	0.110
	Styrene, phm	14	14
	Peak Polymerization		
25	Temperature, °C	87	85
	Peak Polymerization Pressure, kPa (psi)	283 (41)	276 (40)
	Step 5		
30	Butadiene	17.5	21
	Peak Polymerization Temperature, °C	101	107
	Peak Polymerization Pressure, kPa (psi)	373(54)	400(58)
35	Step 6 (Coupling)		
	Vikoflex 7170, phm	0.4	0.4
	Temperature, °C	89	95
	Pressure, kPa (psi)	352 (51)	373 (54)
40	Step 7 (Terminating)		
	Water, phm	0.2	0.2
	Carbon Dioxide, phm	0.1	0.1
45	Temperature, °C	87	87
	Pressure, kPa (psi)	345 (50)	345 (50)
	Step 8 (Stabilizing)		
	Stabilizer Mixture, phm	1.21	1.25
50	Antiblocking agent, phm	0.3	0.3
	Recovered Resin	-	
55	Melt Flow, g/10 min	8.2	8.6

<sup>a</sup>After each addition of monomer, initiator or additive, the feed lines were rinsed with approximately 5-20 phm cyclohexane diluent and cleared with nitrogen.



**Example IX**

Polymers from Examples VII and VIII were blended with Novacor® 555 general purpose polystyrene to produce blends 13', 14', 15', 16', and 17' for evaluation. The resulting blend was extruded on an 8.9 cm diameter extruder and sheet line. All blends were 50:50 by weight except for blend 16', which was a 60:40 copolymer:polystyrene by weight blend. A 0.51 mm die gap opening was used for the operation of a 0.38 mm thick extruded sheet for evaluation.

The extruded sheet samples were evaluated and the results are shown in Table 15. The fold test involves bending the sheet on itself in the machine direction (MD) to induce a transverse directional (TD) break. The sheet is folded at two different rates to allow differentiation between brittle sheet samples. In addition, the behavior of the sheet in response to a tear in machine and transverse directions was determined.

Sheets made from invention blends 13', 14' and 15' all had less blueness than sheets made from comparative blend 16', but more blueness than sheets made from comparative blend 17'.

Sheets made from blend 13' had the highest total energy dart drop value of sheets from this set of blends. Sheets made from blends 14' and 15' had low total energy dart drop values similar to the values of sheets made from blend 16'. Sheets made from comparative blend 16' had the lowest blueness value of any in the group, but had a low total energy dart drop value and brittle fold test. Sheets made from comparative blend 17' had the highest blueness value and better ductility than invention blends 14' and 15'.

TABLE 15

Physical Properties of Polymer Blends With Polystyrene					
Property	Invention Blend 13'	Invention Blend 14'	Invention Blend 15'	Comparative Blend 16'	Comparative Blend 17'
Styrene-butadiene copolymer	13	14	15	16	17
Copolymer: Styrene blend ratio	50:50	50:50	50:50	60:40	50:50
Hunter blueness b	-8.0	-10.0	-10.1	-5.7	-13.5
Haze, %	2.46	2.48	2.97	1.63	3.60
Transmission, %	89.6	89.3	89.2	89.8	88.8
Total energy dart drop, J	3.17	0.80	1.11	0.95	1.61
Transverse direction Fold <sup>a</sup> (fast/slow)	D/D	B/D	B/D	B/B	D/D
Machine direction Tear <sup>a</sup>	D	D	D	D	D
Transverse direction Tear <sup>a</sup>	D	D	D	D	D

<sup>a</sup>D=ductile, B=brittle.

These results show that either of the embodiments of the invention can be used in blends with polymers of styrene to produce resins from which can be made articles having low blueness and other properties comparable to or better than the properties of articles made from blends of polymers of styrene with other monovinylaromatic/conjugated diene copolymers. More particularly, a comparison of the 50:50 (copolymer:styrene) invention blends 13', 14' and 15' with the 60:40 (copolymer:styrene) comparative blend 16' shows that a smaller amount of invention copolymer than comparative copolymer can be used to show comparable improvements in reduction of blueness. Other surprising blend properties are demonstrated by these runs. For example, invention blend 13' which had a total of about 75 wt % total styrene in the blend composition had total energy dart drop of 3.17 J compared to a total energy dart drop of 1.61 J for comparative blend 17' which had only about 70 wt % total styrene in the blend composition.

## Claims

1. A method of preparing a polymer under solution polymerization conditions in a reaction zone comprising:

(a) charging a monovinylaromatic monomer and an initiator in the presence of a randomizer; which is an ether, thioether, or a tertiary amine; thereafter

(b) charging an initiator and a monovinylaromatic monomer; thereafter

(c) charging a mixture of a monovinylaromatic monomer and a conjugated diene monomer; thereafter

(d) optionally charging a mixture of a monovinylaromatic monomer and a conjugated diene monomer; and thereafter

(e) charging the reaction mixture with a coupling agent,

wherein in steps (a) and (b) 0.005 - 1.5 parts by weight per 100 parts by weight of total monomers each of initiator, and 20 - 60 parts by weight per 100 parts by weight of total monomers each of the monovinylaromatic monomer are employed, and

wherein in each of steps (a) to (d) polymerization is conducted until essentially no free monomer is present.

2. The method of claim 1 consisting of steps (a), (b), (c), and (e).

3. The method of claim 1 consisting of steps (a), (b), (c), (d), and (e).

4. The method of any of claims 1 to 3, wherein each said monovinylaromatic monomer contains 8 to 12 carbon atoms and each said conjugated diene contains 4 to 6 carbon atoms, said monomers being added in a ratio of 55 to 95, preferably of 60 to 90, and most preferably of 65 to 85 weight percent, monovinylaromatic monomer to 5 to 45, preferably of 10 to 40, and most preferably of 15 to 35, weight percent conjugated diene monomer, thus giving a resinous tapered block copolymer.

5. The method of any of the preceding claims, wherein the weight ratio of the amounts of said monovinylaromatic monomer and said conjugated diene monomer in each of steps (c) and (d) is within the range of 1:0.63 to 1:2, preferably of 1:0.67 to 1:1.8, and most preferably of 1:0.8 to 1:1.5.

6. The method of any of the preceding claims, wherein said monovinylaromatic monomer charged in step (a) is from 27 to 80, preferably from 40 to 67, and most preferably from 47 to 60, each weight percent of the total monovinylaromatic monomer charged in said method.

7. The method of any of the preceding claims, wherein said monovinylaromatic monomer charged in step (b) is from 5 to 21, preferably from 8 to 19 and most preferably from 11 to 16, weight percent of the total monovinylaromatic monomer charged in said method.

8. The method of any of claims 1, 2 and 4 to 7, wherein said monovinylaromatic monomer charged in step (c) is from 7 to 60, preferably from 27 to 40, and most preferably from 29 to 37, weight percent of the total monovinylaromatic monomer charged in said method.

9. The method of any of claims 1 and 3 to 7, wherein said monovinylaromatic monomer charged in step (c) is from 3 to 19, preferably from 5 to 16, and most preferably from 8 to 13, weight percent of the total monovinylaromatic monomer charged in said method.

10. The method of any of claims 1, 3 to 7 and 9, wherein said monovinylaromatic monomer charged in step (d) is from 13 to 36, preferably from 16 to 32, and most preferably from 20 to 25, weight percent of the total monovinylaromatic monomer charged in said method.

11. The method of any of the preceding claims, wherein said initiator in steps (a) and (b) is charged in amounts effective to provide proportionate amounts of polymeric species in said polymer such that articles made from blends of said polymer with polymers of styrene have improved impact strength.

12. The method of any of the preceding claims, wherein said initiator in steps (a) and (b) is charged in amounts effective to provide proportionate amounts of polymeric species in said polymer such that articles made from blends of said polymer with polymers of styrene have low blueness.

13. The method of any of the preceding claims, wherein said initiator in steps (a) and (b) is charged in amounts effective to provide polymer having a melt flow in the range from 2 to 50 g/10 min.

14. The method of any of the preceding claims, wherein said initiator in step (b) is present in an amount at least equal to the amount of said initiator in step (a), preferably wherein the amount of initiator in step (b) is from 0.25 to 300 times as much, most preferably from 0.3 to 10 times as much, each based on weight, as the amount of initiator in step (a).

15. The method of any of the preceding claims, wherein each said conjugated diene monomer is 1,3-butadiene, each said monovinylaromatic monomer is styrene, said organomonoalkali metal initiator is n-butyllithium, said randomizer is one chosen from tetrahydrofuran and diethyl ether, and said coupling agent is epoxidized vegetable oil.

16. The method of any of the preceding claims, wherein polymerization is carried out in a hydrocarbon diluent;

wherein said polymerization is carried out in a substantial absence of oxygen and water at temperatures ranging from -10 to 150 °C;

wherein, after said coupling agent reacts with products of said polymerization, the system is treated with a terminating agent and a stabilizer; and

wherein, after termination with said terminating agents, any remaining hydrocarbon diluent is flashed off.

17. A polymodal coupled resinous block copolymer of a monovinyl aromatic compound and a conjugated diene, said copolymer having polymer chains which result from coupling a mixture comprising

S-S-B/S-B/S-Li and

S-B/S-B/S-Li

wherein

S = monovinylaromatic block

B = conjugated diene block

B/S = tapered block

Li = living polymer site or coupling site.

18. The copolymer of claim 17 having from 55 to 95, preferably from 60 to 90, and most preferably from 65 to 85, weight percent polymerized monovinylaromatic monomer, based on total weight of monomers in said copolymer.

19. Articles made from the copolymer of claim 17 or 18.

20. Blends of a polymer of styrene and the copolymer of claim 17 or 18.

21. The blend of claim 20, wherein said polymer of styrene is present in an amount in the range from 10 to 70, preferably from 20 to 65, and most preferably from 30 to 60, weight percent, based on total weight of said blend.

22. Articles made from the blend of claim 20 or 21.

## Patentansprüche

1. Verfahren zur Herstellung eines Polymeren unter Bedingungen der Lösungspolymerisation in einer Reaktionszone, umfassend folgende Stufen:

- (a) Zuführen eines monovinylaromatischen Monomeren und eines Initiators in Gegenwart eines Randomisierungsmittels, bei dem es sich um einen Ether, einen Thioether oder ein tertiäres Amin handelt; anschließend
- (b) Zuführen eines Initiators und eines monovinylaromatischen Monomeren; anschließend

- (c) Zuführen eines Gemisches aus einem monovinylaromatischen Monomeren und einem konjugierten Dienmonomeren; anschließend  
 (d) gegebenenfalls Zuführen eines Gemisches aus einem monovinylaromatischen Monomeren und einem konjugierten Dienmonomeren; und anschließend  
 5 (e) Versetzen des Reaktionsgemisches mit einem Kupplungsmittel;

wobei in den Stufen (a) und (b) 0,005-1,5 Gewichtsteile der einzelnen Initiatoren pro 100 Gewichtsteile der gesamten Monomeren und 20-60 Gewichtsteile der einzelnen monovinylaromatischen Monomeren pro 100 Gewichtsteile der gesamten Monomeren verwendet werden und wobei in den Stufen (a) bis (d) die Polymerisation  
 10 jeweils durchgeführt wird, bis im wesentlichen kein freies Monomeres mehr vorhanden ist.

2. Verfahren nach Anspruch 1, bestehend aus den Stufen (a), (b), (c) und (e).
3. Verfahren nach Anspruch 1, bestehend aus den Stufen (a), (b), (c), (d) und (e).
- 15 4. Verfahren nach einem der Ansprüche 1 bis 3, wobei jedes monovinylaromatische Monomere 8 bis 12 Kohlenstoffatome und jedes konjugierte Dien 4 bis 6 Kohlenstoffatome aufweist, wobei die Monomeren im einem Verhältnis von 55 bis 95, vorzugsweise 60 bis 90 und insbesondere 65 bis 85 Gew.-% des monovinylaromatischen Monomeren zu 5 bis 45, vorzugsweise 10 bis 40 und insbesondere 15 bis 35 Gew.-% des konjugierten Dienmonomeren  
 20 zugesetzt werden, wodurch man ein harzartiges konisches Blockcopolymeres erhält.
5. Verfahren nach einem der vorstehenden Ansprüche, wobei das Gewichtsverhältnis der Mengen des monovinylaromatischen Monomeren und des konjugierten Dienmonomeren in den einzelnen Stufen (c) und (d) im Bereich von 1:0,63 bis 1:2, vorzugsweise von 1:0,67 bis 1:1,8 und insbesondere von 1:0,8 bis 1:1,5 liegt.
- 25 6. Verfahren nach einem der vorstehenden Ansprüche, wobei das in Stufe (a) vorgelegte monovinylaromatische Monomere 27 bis 80, vorzugsweise 40 bis 67 und insbesondere 47 bis 60 Gew.-% des gesamten, beim Verfahren vorgelegten monovinylaromatischen Monomeren ausmacht.
- 30 7. Verfahren nach einem der vorstehenden Ansprüche, wobei das in Stufe (b) vorgelegte monovinylaromatische Monomere 5 bis 21, vorzugsweise 8 bis 19 und insbesondere 11 bis 17 Gew.-% des gesamten, beim Verfahren vorgelegten monovinylaromatischen Monomeren ausmacht.
- 35 8. Verfahren nach einem der vorstehenden Ansprüche 1, 2 und 4 bis 7, wobei das in Stufe (c) vorgelegte monovinylaromatische Monomere 7 bis 60, vorzugsweise 27 bis 40 und insbesondere 29 bis 37 Gew.-% des gesamten, beim Verfahren vorgelegten monovinylaromatischen Monomeren ausmacht.
- 40 9. Verfahren nach einem der Ansprüche 1 und 3 bis 7, wobei das in Stufe (c) vorgelegte monovinylaromatische Monomere in einer Menge von 3 bis 19, vorzugsweise 5 bis 16 und insbesondere 8 bis 13 Gew.-% des gesamten, beim Verfahren vorgelegten monovinylaromatischen Monomeren ausmacht.
- 45 10. Verfahren nach einem der Ansprüche 1, 3 bis 7 und 9, wobei das in Stufe (d) vorgelegte monovinylaromatische Monomere 13 bis 36, vorzugsweise 16 bis 32 und insbesondere 20 bis 25 Gew.-% des gesamten, beim Verfahren vorgelegten monovinylaromatischen Monomeren ausmacht.
- 50 11. Verfahren nach einem der vorstehenden Ansprüche, wobei der Initiator in den Stufen (a) und (b) in Mengen vorgelegt wird, die bewirken, daß proportionale Mengen der polymeren Spezies im Polymeren bereitgestellt werden, so daß aus den Gemischen des Polymeren mit Polymeren von Styrol hergestellte Gegenstände eine verbesserte Schlagzähigkeit aufweisen.
- 55 12. Verfahren nach einem der vorstehenden Ansprüche, wobei der Initiator in den Stufen (a) und (b) in Mengen vorgelegt wird, die bewirken, daß proportionale Mengen an polymerer Spezies im Polymeren bereitgestellt werden, so daß aus Gemischen des Polymeren mit Polymeren von Styrol gefertigte Gegenstände eine geringe Blau aufweisen.
13. Verfahren nach einem der vorstehenden Ansprüche, wobei der Initiator in den Stufen (a) und (b) in Mengen vorgelegt wird, die bewirken, daß ein Polymeres mit einem Schmelzindex von 2 bis 50 g/10 min bereitgestellt wird.

14. Verfahren nach einem der vorstehenden Ansprüche, wobei der Initiator in Stufe (b) in einer Menge vorliegt, die mindestens der Menge des Initiators in Stufe (a) entspricht, wobei vorzugsweise die Menge des Initiators in Stufe (b) das 0,25- bis 300-fache und vorzugsweise das 0,3-bis 10-fache (jeweils bezogen auf das Gewicht) der Menge des Initiators in Stufe (a) beträgt.

15. Verfahren nach einem der vorstehenden Ansprüche, wobei es sich bei den einzelnen konjugierten Dienmonomeren um 1,3-Butadien, bei den einzelnen monovinylaromatischen Monomeren um Styrol, beim Organomonoalkalimetall-Initiator um n-Butyllithium, beim Randomisierungsmittel um Tetrahydrofuren oder Diethylether und beim Kupplungsmittel um epoxidiertes pflanzliches Öl handelt.

16. Verfahren nach einem der vorstehenden Ansprüche, wobei die Polymerisation in einem Kohlenwasserstoff-Verdünnungsmittel durchgeführt wird;

wobei die Polymerisation im wesentlichen in Abwesenheit von Sauerstoff und Wasser bei Temperaturen im Bereich von -10 bis 150°C durchgeführt wird;

wobei das System, nachdem das Kupplungsmittel mit den Produkten der Polymerisation reagiert hat, mit einem Terminationsmittel und einem Stabilisator behandelt wird; und

wobei nach Termination mit den Terminationsmitteln etwaiges verbleibendes Kohlenwasserstoff-Verdünnungsmittel durch Flash-Verdampfung entfernt wird.

17. Polymodales, gekuppeltes, harzartiges Blockcopolymeres aus einer monovinylaromatischen Verbindung und einem konjugierten Dien, wobei das Copolymeres Polymerketten aufweist, die sich aus der Kupplung eines Gemisches mit folgenden Bestandteilen ergibt:

S-S-B/S-B/S-Li und

S-B/S-B/S-Li

worin

S = monovinylaromatischer Block

B = konjugierter Dienblock

B/S = konischer Block

Li = lebende Polymerstelle oder Kupplungsstelle

18. Copolymeres nach Anspruch 17, mit einem Anteil von 55 bis 95, vorzugsweise 60 bis 90 und insbesondere 65 bis 85 Gew.-% an polymerisiertem monovinylaromatischem Monomeren, bezogen auf das Gesamtgewicht der Monomeren im Copolymeren.

19. Gegenstände, gefertigt aus dem Copolymeren nach Anspruch 17 oder 18.

20. Gemische eines Polymeren von Styrol und dem Copolymeren nach Anspruch 17 oder 18.

21. Gemisch nach Anspruch 20, wobei das Polymere von Styrol in einem Anteil von 10 bis 70, vorzugsweise 20 bis 65 und insbesondere 30 bis 60 Gew.-%, bezogen auf das Gesamtgewicht des Gemisches vorhanden ist.

22. Gegenstände, gefertigt aus dem Gemisch nach Anspruch 20 oder 21.

## Revendications

1. Un procédé de préparation d'un polymère sous des conditions de polymérisation en solution dans une zone de réaction comprenant les étapes consistant à :

(a) charger un monomère monovinylaromatique et un amorceur en présence d'un agent de randomisation qui est un éther, un thioéther ou une tertio-amine ; puis

(b) charger un amorceur et un monomère monovinylaromatique ; puis

(c) charger un mélange d'un monomère monovinylaromatique et d'un monomère diénique conjugué ; puis

(d) charger éventuellement un mélange d'un monomère monovinylaromatique et d'un monomère diénique conjugué ; et ensuite

(e) charger le mélange de réaction avec un agent de couplage,

dans lequel dans les étapes (a) et (b), on utilise 0,005 à 1,5 partie en poids pour 100 parties en poids des monomères totaux chacun de l'amorceur et de 20 à 60 parties en poids pour 100 parties en poids des monomères totaux chacun du monomère monovinylaromatique et dans lequel dans chacune des étapes (a) à (d), la polymérisation est mise en oeuvre dans des conditions dans lesquelles sensiblement aucun monomère libre est présent.

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2. Le procédé selon la revendication 1 qui consiste dans la mise en oeuvre des étapes (a), (b), (c) et (e).

3. Le procédé selon la revendication 1 qui consiste dans la mise en oeuvre des étapes (a), (b), (c), (d) et (e).

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4. Le procédé selon l'une quelconque des revendications 1 à 3, dans lequel chaque monomère monovinylaromatique renferme 8 à 12 atomes de carbone et chaque diène conjugué renferme 4 à 6 atomes de carbone, lesdits monomères étant ajoutés dans un rapport de 55 à 95, de préférence de 60 à 90, et mieux encore de 65 à 85 en poids de monomère monovinylaromatique pour 5 à 45, de préférence de 10 à 40, et mieux encore de 15 à 35 % en poids de monomère diénique conjugué, ce qui donne un copolymère à séquences progressives résineux.

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5. Le procédé selon l'une quelconque des revendications précédentes, dans lequel le rapport pondéral des quantités dudit monomère monovinylaromatique et dudit monomère diénique conjugué dans chacune des étapes (c) et (d), est compris dans la gamme de 1:0,63 à 1:2, de préférence de 1:0,67 à 1:1,8 et mieux encore de 1:0,8 à 1:1,5.

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6. Le procédé selon l'une quelconque des revendications précédentes, dans lequel ledit monomère monovinylaromatique chargé dans l'étape (a) représente de 27 à 80, de préférence de 40 à 67, et mieux encore de 47 à 60 % en poids du monomère monovinylaromatique total chargé dans ledit procédé.

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7. Le procédé selon l'une quelconque des revendications précédentes, dans lequel ledit monomère monovinylaromatique chargé dans l'étape (b) représente de 5 à 21, de préférence de 8 à 19 et mieux encore de 11 à 16 % en poids du monomère monovinylaromatique total chargé dans ledit procédé.

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8. Le procédé selon l'une quelconque des revendications 1, 2 et 4 à 7, dans lequel ledit monomère monovinylaromatique chargé dans l'étape (c) représente de 7 à 60, de préférence de 27 à 40, et mieux encore de 29 à 37 % en poids du monomère monovinylaromatique total chargé dans ledit procédé.

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9. Le procédé selon l'une quelconque des revendications 1 et 3 à 7, dans lequel ledit monomère monovinylaromatique chargé dans l'étape (c) représente de 3 à 19, de préférence de 5 à 16, et mieux encore de 8 à 13 % en poids du monomère monovinylaromatique total chargé dans ledit procédé.

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11. Le procédé selon l'une quelconque des revendications précédentes, dans lequel ledit amorceur dans les étapes (a) et (b) est chargé selon des quantités efficaces pour fournir des quantités proportionnées d'espèces polymères dans ledit polymère de sorte que les articles obtenus à partir des mélanges dudit polymère avec des polymères de styrène ont une résistance au choc améliorée.

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12. Le procédé selon l'une quelconque des revendications précédentes dans lequel ledit amorceur dans les étapes (a) et (b) est chargé selon des quantités efficaces pour fournir des quantités proportionnées d'espèces polymères dans ledit polymère de sorte que les articles obtenus à partir des mélanges dudit polymère avec des polymères de styrène ont un faible bleuté.

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13. Le procédé selon l'une quelconque des revendications précédentes, dans lequel ledit amorceur dans les étapes (a) et (b) est chargé selon des quantités efficaces pour fournir un polymère ayant un écoulement à l'état fondu dans la gamme de 2 à 50 g/10 min.

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14. Le procédé selon l'une quelconque des revendications précédentes, dans lequel ledit amorceur dans l'étape (b) est présent selon une quantité au moins égale à la quantité dudit amorceur dans l'étape (a), de préférence dans lequel la quantité de l'amorceur dans l'étape (b) est de 0,25 à 300 fois ou mieux encore de 0,3 à 10 fois sur une base pondérale la quantité de l'amorceur dans l'étape (a).

15. Le procédé selon l'une quelconque des revendications précédentes, dans lequel chaque monomère diénique conjugué est du 1,3-butadiène, chaque monomère monovinylaromatique est du styrène, ledit amorceur de métal organomonoalcalin est le n-butyllithium, ledit agent de randomisation est choisi parmi le tétrahydrofuranne et l'éther diéthylique, et ledit agent de couplage est une huile végétale époxydée.

16. Le procédé selon l'une quelconque des revendications précédentes, dans lequel la polymérisation est mise en oeuvre dans un diluant hydrocarboné, dans lequel ladite polymérisation est mise en oeuvre en l'absence sensible d'oxygène et d'eau à une température comprise de -10 à 150°C, dans lequel après que ledit agent de couplage a réagi avec les produits de ladite polymérisation, le système est traité avec un agent de terminaison et un stabilisant et dans lequel après la terminaison avec lesdits agents de terminaison, tout diluant hydrocarboné restant est séparé par distillation.

17. Un copolymère séquencé résineux à couplage polymodal d'un composé monovinylaromatique et d'un diène conjugué, ledit copolymère ayant des chaînes de polymère qui résultent d'un couplage d'un mélange comprenant

S-S-B/S-B/S-Li et

S-B/S-B/S-Li

où

S = séquence monovinylaromatique

B = séquence de diène conjugué

B/S = séquence progressive

Li = site de polymère vivant ou site de couplage

18. Le copolymère selon la revendication 17, ayant de 55 à 95, de préférence de 60 à 90, et mieux encore de 65 à 85 % en poids de monomère monovinylaromatique polymérisé, calculé sur le poids total des monomères dans ledit copolymère.

19. Articles obtenus à partir du copolymère de la revendication 17 ou 18.

20. Mélanges d'un polymère de styrène et du copolymère de la revendication 17 ou 18.

21. Le mélange de la revendication 20, dans lequel ledit polymère de styrène est présent selon une quantité comprise dans la gamme de 10 à 70, de préférence de 20 à 65 et mieux encore de 30 à 60 % en poids, calculé sur le poids total dudit mélange.

22. Articles réalisés à partir du mélange de la revendication 20 ou 21.

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